

請依照題號次序作答

I. (50%) Simple choice. There is only one correct answer in each question.

- Which compound is the major component in natural gas? (A) methane (B) nitrogen (C) acetylene (D) ethene
- Which functional group is **not** contained in cysteine $\text{HSCH}_2\text{CH}(\text{NH}_2)\text{COOH}$? (A) amino (B) carboxylic acid (C) hydroxyl (D) thiol
- Which compound reacts rapidly with Br_2 and discharge the brown color? (A) hexane (B) 3-hexene (C) 3-hexanone (D) 3-hexanol
- How many chiral carbon(s) in a molecule of 2,3-dimethylbutane? (A) 0 (B) 1 (C) 2 (D) 3
- The triple bond of an alkyne $\text{RC}\equiv\text{CR}$ is composed of (A) three σ bonds (B) three π bonds (C) one σ bond and two π bonds (D) two σ bonds and one π bond
- Which compound is the strongest acid? (A) HCN (B) H_2CO_3 (C) $\text{CH}_3\text{CO}_2\text{H}$ (D) $\text{CF}_3\text{CO}_2\text{H}$
- Which reagent can convert 2-methyl-1-propene $(\text{CH}_3)_2\text{C}=\text{CH}_2$ to 1,1-dimethylethanol $(\text{CH}_3)_3\text{COH}$? (A) O_3 (B) H_2O_2 (C) 6 M $\text{NaOH}/\text{H}_2\text{O}$ (D) 6 M $\text{H}_2\text{SO}_4/\text{H}_2\text{O}$
- Which compound reacts with Tollens reagent $\text{Ag}(\text{NH}_3)_2^+$ to give a positive silver mirror test? (A) pentanal (B) pentanol (C) 2-pentene (D) 2-pentanone
- Which reaction is good to prepare *tert*-butyl methyl ether $(\text{CH}_3)_3\text{COCH}_3$? (A) $(\text{CH}_3)_3\text{CO}^- \text{K}^+ + \text{CH}_3\text{I}$ (B) $(\text{CH}_3)_3\text{C}^- \text{K}^+ + \text{CH}_3\text{OH}$ (C) $(\text{CH}_3)_3\text{Cl} + \text{CH}_3\text{O}^- \text{Na}^+$ (D) $(\text{CH}_3)_3\text{COH} + \text{CH}_3\text{O}^- \text{Na}^+$
- Which compound reacts most rapidly with Br_2 in the presence of NaHCO_3 ? (A) chlorobenzene ClC_6H_5 (B) toluene $\text{C}_6\text{H}_5\text{CH}_3$ (C) aniline $\text{C}_6\text{H}_5\text{NH}_2$ (D) nitrobenzene $\text{C}_6\text{H}_5\text{NO}_2$
- Which compound has the highest boiling point? (A) pentane (B) diethyl ether (C) 2-butanone (D) 2-butanol
- Which compound is a tertiary amine? (A) butylamine (B) diethylamine (C) 2,2-dimethylethylamine (D) *N,N*-dimethylethylamine
- Which compound is insoluble in water but dissolves in aqueous NaHCO_3 solution with a release of CO_2 gas? (A) aldehyde (B) ester (C) carboxylic acid (D) ketone
- The relationship between acetone and 2-hydroxy-1-propene is (A) tautomers (B) enantiomers (C) diastereomers (D) conformational isomers
- For D-glyceraldehyde $\text{HOH}_2\text{C}-\text{CH}(\text{OH})-\text{CHO}$ (A) it is achiral (B) it is a planar molecule (C) it has (*R*)-configuration (D) 1,3-dihydroxyacetone is its configurational isomer
- Peptide bonds are (A) ester bonds (B) amide bonds (C) hydrogen bonds (D) acetal bonds
- Compound $\text{CH}_3(\text{CH}_2)_{16}\text{CO}_2\text{H}$ is (A) a bile acid (B) a fatty acid (C) a nucleic acid (D) an amino acid
- Nature rubber $[\text{CH}_2\text{CH}=\text{C}(\text{Me})\text{CH}_2]_n$ is formed by polymerization from monomers of (A) 2-methyl-2-butene (B) 3-methyl-2-butene (C) 2-methyl-1,3-butadiene (D) 3-methyl-1,3-butadiene
- Which IR absorption corresponds to the carbonyl group of acetone $(\text{CH}_3)_2\text{C}=\text{O}$? (A) 1200 cm^{-1} (B) 1710 cm^{-1} (C) 2100 cm^{-1} (D) 3300 cm^{-1}
- The reaction of (*E*)-2-butene with *m*-chloroperoxybenzoic acid will give (A) *meso*-2,3-butanediol (B) racemic 2,3-butanediol (C) *cis*-1,2-dimethyloxirane (D) *trans*-1,2-dimethyloxirane
- The reaction of 1,2-epoxycyclohexane with phenylmagnesium bromide $\text{C}_6\text{H}_5\text{MgBr}$, after aqueous workup, will give (A) 1-phenylcyclohexanol (B) *trans*-2-phenylcyclohexanol (C) *cis*-2-phenylcyclohexanol (D) cyclohexyl phenyl ether

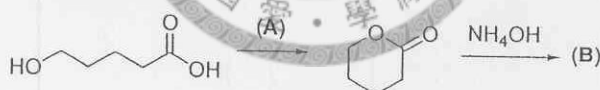
22. Which compound is the most stable? (A) 1,4-pentadiene (B) *trans*-1,3-pentadiene (C) *trans*-1,4-hexadiene (D) *cis*-1,4-hexadiene
23. Which reagent is used to convert methyl benzoate $\text{C}_6\text{H}_5\text{COOCH}_3$ to benzyl alcohol and methanol? (A) NaBH_4 (B) NaOH (C) LiAlH_4 (D) CrO_3
24. Which reagent is used to convert benzene to bromobenzene? (A) HBr (B) PBr_3 (C) Br_2/NaOH (D) $\text{Br}_2/\text{FeBr}_3$
25. Which product is obtained from the Claisen condensation of ethyl acetate $\text{CH}_3\text{CO}_2\text{C}_2\text{H}_5$? (A) ethyl acetoacetate $\text{CH}_3\text{COCH}_2\text{CO}_2\text{C}_2\text{H}_5$ (B) ethyl butanoate $\text{CH}_3\text{CH}_2\text{CH}_2\text{CO}_2\text{C}_2\text{H}_5$ (C) ethyl 3-hydroxybutanoate $\text{CH}_3\text{CH}(\text{OH})\text{CH}_2\text{CO}_2\text{C}_2\text{H}_5$ (D) ethyl 2-butenate $\text{CH}_3\text{CH}=\text{CHCO}_2\text{C}_2\text{H}_5$

II. (10%) Draw Newman projections for two staggered and two eclipsed conformations of 1,2-dichloroethane. Indicate which conformation is the most stable, and which conformation is the least stable.

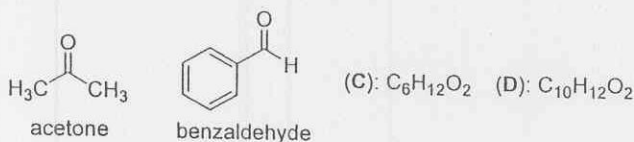
III. (10%) Propose a mechanism for the addition of HI to 1-methyl-1-cyclopentene to give 1-iodo-1-methylcyclopentane. Why this reaction disfavors the formation of 2-iodo-1-methylcyclopentane?

IV. (10%) Arrange the following compounds in order of increasing acidity: phenol $\text{C}_6\text{H}_5\text{OH}$, benzyl alcohol $\text{C}_6\text{H}_5\text{CH}_2\text{OH}$ and 4-nitrophenol $\text{O}_2\text{NC}_6\text{H}_4\text{OH}$. Explain this trend of acidity.

V. (10%) Treatment of a carboxylic acid with an alcohol in the presence of an acid catalyst to give an ester is known as Fischer esterification. Give reagent (A) and product (B) to complete the following reactions.



VI. (10%) Aldol reactions are the condensation reactions between two aldehydes or ketones. The base-catalyzed aldol reactions between acetone and benzaldehyde may give two products (C) and (D). Dehydration of compounds (C) and (D) gives compounds (E) and (F), respectively. Show the structures of compounds (C), (D), (E) and (F).



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