國立台灣大學九十四學年度碩士班招生考試試題

科目:有機化學(A)

題號: 69

頁之第 頁

單選題 (60%)(不需抄題但請標明題號並依序作答)

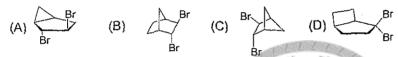
What is the best sequence of reactions to synthesize m-nitrophenol?

(A)
$$OH - HNO_3 - POH - HNO_3 - POH - HNO_3 - POH - P$$

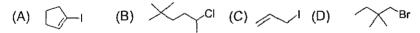
Which of the following is the enol tautomer of 2-butanone?

$$(A) \ \, \bigcap_{\text{CH}_2 = \text{CH-CH-CH}_3}^{\text{OH}} \ \, (B) \ \, \bigcap_{\text{II}}^{\text{II}} \ \, (C) \ \, \bigcap_{\text{CH}_3 - \text{C} = \text{CH-CH}_3}^{\text{OH}} \ \, (D) \ \, \bigcap_{\text{CH}_3 \text{CH}_2 \text{CH}(\text{OH})\text{CHO}}^{\text{OH}} \ \, (D) \ \, \bigcap_{\text{CH}_3 \text{CH}_2 \text{CH}(\text{OH})\text{CHO}}^{\text{CH}_3 \text{CH}_2 \text{CH}(\text{OH})\text{CHO}}^{\text{CH}_3 \text{CH}_2 \text{CH}_3} \ \, (D) \ \, \bigcap_{\text{CH}_3 \text{CH}_3 \text{CH}_3$$

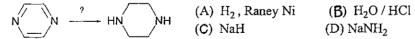
- How many peaks would you expect to observe in the ¹H NMR spectrum for 1,3-diiodo-2,2-dimethylpropane? (A) 2 (B) 3 (C) 4 (D) 10
- Which of the following compounds is the strongest base?
 - (A) CH_3NH_2 (B) Pyridine (C) $(CH_3)_4N^{\dagger}Br$ (D) $(CH_3CH_2)_2NH$
- Which compound is the product when bicyclo[2,2,1]-2-heptene reacts with Br₂/CCl₄?



- Which of the following is the correct structure of allyl chloride?
 - (A) CH3CH=CHCl (B) ClCH2CH=CH2 (C) CHCl=CHCH3 (D) CH2=CHCHCl2
- 7. What would be the reduction products of C₆H₅CH=CHCO₂CH₂CH₃ with LiAlH₄ after workup?
 - (A) C₆H₅CH=CHCH₂OH + CH₃CH₂OH
- (B) C₆H₅CH₂CH₂OH + CH₃CH₂OH
- (C) $C_0H_5CHCHO + OHCCH_2CH_2OCH_2CH_3$ (D) $C_0H_5CH=CHCH_2OCH_2CH_3 + CH_3OH$
- Which of the following compounds has the highest boiling point?
 - (A) o-xylene (B) benzyl alcohol (C) benzene (D) diethyl ether
- Which of the following reaction will yield CH3CHDCH3 as the product?
 - (A) $CH_3CH_2CH_2MgBr + D_2O \rightarrow ?$
- (B) $CH_3CHDCH_2MgBr + D_2O \rightarrow ?$
 - (C) $(CH_3)CHLi + D_2O \rightarrow ?$
- (D) $CH_3CHDCH_2OH + D_2SO_4 \rightarrow ?$
- 10. Which of the following alkyl halides has the fastest substitution rate with KCN in acetone at room temperature?



- 11. Which of the following is a polar aprotic solvent?
 - (B) (CH₃)₂S=O (C) CH₃CH₂OH (D) CH₃COOH
- 12. What set of reagents is required for the following reaction?



- 13. A pentapeptide contains the amino acids: Gly, Leu, Ala, Phe(2). Partial hydrolysis of the peptide gives fragments: Leu-Gly-Phe, Phe-Ala and Gly-Phe-Phe. What is the structure of peptide?
 - (A) Leu-Gly-Phe-Phe-Ala (B) Phe-Phe-Ala-Leu-Gly (C) Gly-Leu-Ala-Phe-Phe (D) Leu-Gly-Phe-Ala-Phe
- 14. What would be the Major product of the following reaction?

- 15. What would be produced from the reaction of D-glucose with Fehling's solution?
 - (A) D-glucital + Cu₂O
- (B) D-mannaric acid + Cu₂O
- (C) L-gluconic acid + Cu(s)
- (D) D-gluconic acid + Cu₂O
- 16. Which of the following amino acid contains an imidazole ring?
 - (A) histidine (B) glycine (C) aspartic acid (D) serine

國立台灣大學九十四學年度碩士班招生考試試題

科目:有機化學(A)

題號: 69

共 2 頁之第 2

17. What product(s) would be obtained from the reaction of cyclopentene with O₃ followed by the treatment of zinc metal in CH3COOH?

(A) OHCCH2CH2CHO (B) OHCCH2CH2CH2CHO (C) CH3CH2CHO + CH3CHO (D) 2 CH3CHO

18. Which of the following compounds will give a Positive "Tollens' test"?

(A) $C_0H_5CH_2COOH$ (B) $C_0H_5COCH_2CH_3$ (C) $C_0H_5CH(OCH_2CH_3)_2$ (D) $p-CH_3OC_0H_4CHO$

19. What is the final product for the following transformation?

20. What would be the product of the reaction of C₆H₅NCS with CH₃CH₂CH₂NH₂?

問答題

- -. t-Butyl methyl ether is an additive for unleaded gasoline. How to prepare this compound from isobutylene?
- ... Maltose is a disaccharide composed of one D-glucose unit joined to a second D-glucose unit through a α-1,4' linkage. Please draw the structure of: (A) \alpha-D-glucopyranose in a chair conformation. (B) Maltose. (8 %)
- Ξ. Please explain each of the following terms. (12%)
 - (a) Aldol reaction (b) meso compound (c) isoelectric point (d) t-butyl radical
- III. Please give the structure of 2'-deoxyadenosine 5'-phosphate. (5%)
- Provide the organic product(s) for the following reactions: (10%)

(A)
$$\frac{1. \text{ CH}_3\text{ONa}}{2. o\text{-FC}_6\text{H}_5\text{CH}_2\text{Br}}?$$
(B) $(\text{CH}_3)_3\text{CBr} + \text{CH}_3\text{CH}_2\text{OK} \xrightarrow{\text{THF}}?$
(C)
$$\frac{1. \text{CH}_3\text{ONa}}{2. o\text{-FC}_6\text{H}_5\text{CH}_2\text{Br}}?$$
(B) $(\text{CH}_3)_3\text{CBr} + \text{CH}_3\text{CH}_2\text{OK} \xrightarrow{\text{THF}}?$

$$(C)$$
 \bigcirc COOH + \bigcirc NH $\stackrel{25^{\circ}C}{\longrightarrow}$