

1. Please define the activity and activity coefficient for nonideal solutions. (10%)
2. Please derive the heat capacity at constant P, $C_p = C_v + \frac{TV\alpha^2}{\kappa}$, where α is the coefficient of thermal expansion and κ is the compressibility. (15%)
3. Please define the second law of thermodynamics in detail. (10%)
4. Please define the ionic strength and describe how it relates to the Debye-Hückel limiting law. (10%)
5. Please describe the van der Waals equation in detail. How is it different from the ideal gas law? (10%)
6. What is the surface tension? How to measure it? Please give a governor equation for your method. (15%)
7. What is the osmotic pressure? How to measure it? Please give a governor equation for your method. (15%)
8. Please indicate the difference between the Gibbs free energy and Helmholtz free energy. (5%)
9. Please describe the Boltzmann distribution law. How does it relate to the entropy. (10%)